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Note : # Neutrons = Mass Number – Atomic Number. # Neutrons of Nitrogen-14 = 14 - 7 = 7. # Neutrons of Nitrogen-13 = 13 - 7 = 6. c) The number of electrons in a neutral atom always equals the atomic number. The atomic number of both isotopes is 7, therefore the number of electrons in both isotopes is also 7.

Organic Chemistry, 5th Edition Chapter 1 - Section 1.1 ...

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